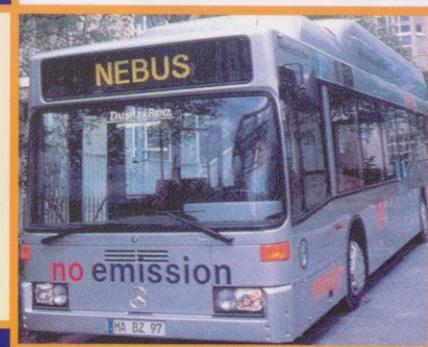
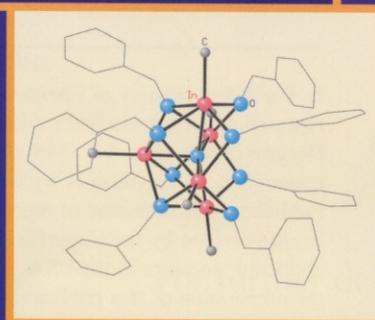
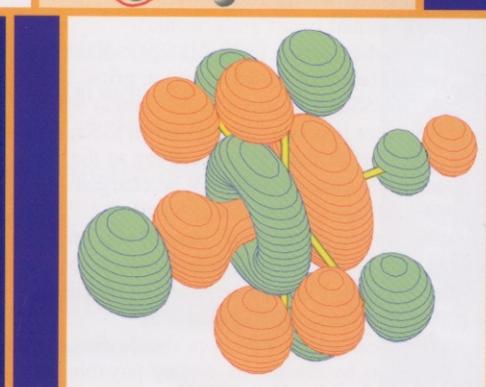
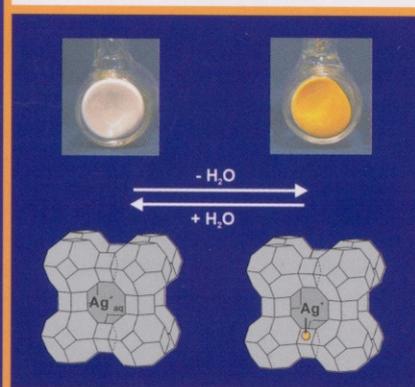
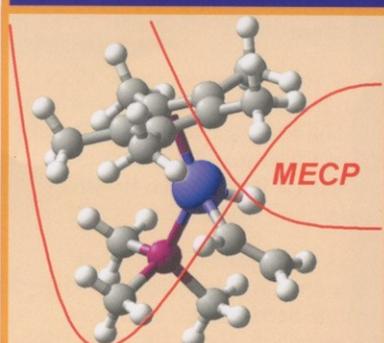
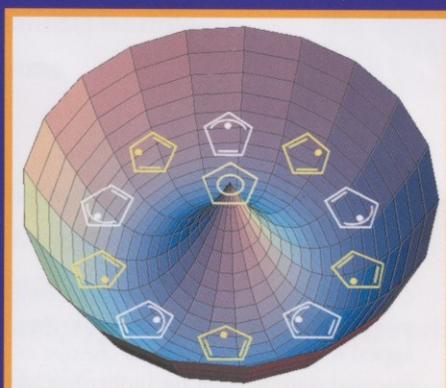


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The electronic structure of Cu⁺, Ag⁺, and Au⁺ zeolites



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A variety of procedures have been used to prepare d¹⁰-zeolite materials. The electronic structure of these materials can be regarded to a first approximation as a superposition of the framework, of the charge compensating ions, of solvent molecules and of guest species. Zeolite oxygen to d¹⁰-ion charge transfer transitions dominate the electronic spectra if the ions coordinate to the zeolite oxygens. Specific

coordination sites can influence the energy and the intensity of these transitions remarkably. Intra guest transitions dominate in quantum dot materials, as discussed in detail for luminescent Ag₂S zeolite A. The zeolite is not needed for the photocatalytic water oxidation on Ag⁺/AgCl photoanodes with visible light. It can, however, be used to increase the active surface area substantially.

Gion Calzaferri (left) received his PhD degree at the University of Fribourg, Switzerland, 1971. He moved to the University of Berne, after a postdoc period at the University of Basel and physical-chemical research in connection with pharmacy at Ciba-Geigy Basel, Switzerland. He has been professor of physical chemistry in the Department of Chemistry and Biochemistry since 1988. His current research interests concern supramolecularly organized molecules, clusters and complexes in zeolite cavities, dye-zeolite materials as photonic antenna systems, and photochemical transformation and storage of solar energy.

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